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Characterization of THB1, a *Chlamydomonas reinhardtii* Truncated Hemoglobin: Linkage to Nitrogen Metabolism and Identification of Lysine as the Distal Heme Ligand

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**ABSTRACT:** The nuclear genome of the model organism *Chlamydomonas reinhardtii* contains genes for a dozen hemoglobins of the truncated lineage. Of those, THB1 is known to be expressed, but the product and its function have not yet been characterized. We present mutagenesis, optical, and nuclear magnetic resonance data for the recombinant protein and show that at pH near neutral in the absence of added ligand, THB1 coordinates the heme iron with the canonical proximal histidine and a distal lysine. In the cyanomet state, THB1 is structurally similar to other known truncated hemoglobins, particularly the heme domain of *Chlamydomonas eugametos* Li637, a light-induced chloroplastic hemoglobin. Recombinant THB1 is capable of binding nitric oxide (NO) either in the ferric or ferrous state and has efficient NO dioxygenase activity. By using different *C. reinhardtii* strains and growth conditions, we demonstrate that the expression of THB1 is under the control of the NIT2 regulatory gene and that the hemoglobin is linked to the nitrogen assimilation pathway.

In 1992, Potts and co-workers discovered in the cyanobacterium *Nostoc commune* a gene encoding a product resembling protozoan and mammalian myoglobins. It was remarkable that a protein commonly associated with dioxygen transport was found in cyanobacteria and intriguing that the gene was located within a nitrogen fixation operon. Subsequent investigations have provided evidence that *N. commune* "cyanoglobin" belongs to a distinct and well-represented lineage of the hemoglobin (Hb) superfamily, which has been given the general name of truncated hemoglobins (TrHbs) in reference to their primary structures containing ~30 fewer amino acids than myoglobin (Mb). *N. commune* cyanoglobin is thought to be involved in dioxygen scavenging, presumably to protect the nitrogenase from oxidative damage.

Shortly after the discovery of cyanoglobin, two light-induced TrHb genes, *Li410* and *Li637*, were found in the unicellular photosynthetic eukaryotic alga *Chlamydomonas eugametos* (moewusii). Crystallization and subsequent structure determination of the globin domain of the *Li637* gene product (hereafter CtrHb) provided the first glimpse of a TrHb fold. The heme group is buried in the protein, which adopts a two-on-two helical topology resulting from the shortened sequence. *Li637*, a protein targeted to the chloroplast, was one of the first TrHbs to be studied in vivo. Protection against oxidative damage was also proposed as a function for this Hb.

Genes encoding proteins containing one or more Hb domains are present in virtually all forms of life and occur in three distinct lineages, including the TrHb family. Recent genome analyses show a preponderance of TrHbs in cyanobacteria and green algae. The phylogenetic data support the rise of TrHbs soon after the appearance of life on this planet, emerging perhaps 3 billion years ago. During this early phase of life on Earth, atmospheric dioxygen levels were low (10−6 atm). Elevation of the dioxygen level resulted from subsequent geological upheaval and the evolution of photosynthesis roughly 2 billion years ago. Given that the TrHb scaffold was formed long before the development of our current aerobic environment, an attractive original role for these proteins is dioxygen detoxification, as put forth for the modern *N. commune* and *C. eugametos* proteins.

The search for globin physiological functions in unicellular photosynthetic organisms is further guided by the observation that many Hbs (including members of the TrHb family) are capable of reacting with nitrogen–oxygen compounds such as nitric oxide (NO). The reactions include oxidation, e.g., generation of nitrate from nitric oxide by oxy Hb; reduction, e.g., formation of nitric oxide...
from nitrite by deoxy Hb; and isomerization, e.g., conversion of peroxynitrite to nitrate by ferric (met) Hb, each process depending on the redox and ligation state of the protein. Such chemical versatility can be attained by alterations of the heme environment without disruption of the fold. Reactive nitrogen species (RNS) are present in and around photosynthetic algae and cyanobacteria as products of enzymes involved in nitrite or nitrate reduction; they also are produced in the diffusion-limited reaction of NO and superoxide, a reactive oxygen species (ROS). Thus, deoxygenation and RNS regulation are conceivable functions for TrHbs in these organisms.

Among RNS, NO* is a focus of Hb research as these proteins are generally capable of NO* dioxygenation. The overall NO* dioxygenase (NOD) reaction can be summarized as:

\[
\text{Hb-Fe(II)} + O_2 \rightleftharpoons \text{Hb-Fe(III)} - O_2
\]

1

\[
\text{Hb-Fe(III)} - O_2^− + \text{NO}^* \rightarrow \left[\text{Hb-Fe(III)} - \text{OONO}^−\right]
\]

2

\[
\left[\text{Hb-Fe(III)} - \text{OONO}^−\right] \rightarrow \text{Hb-Fe(III)} + \text{NO}_2^−
\]

3

\[
\text{Hb-Fe(III)} + e^- \rightarrow \text{Hb-Fe(II)}
\]

4

To undergo multiple turnovers, the protein must be reduced to the O₂-binding competent ferrous state (step 4). Unlike the flavohemoglobinins, which contain a flavin adenine dinucleotide (FAD) binding domain that allows for efficient heme redox cycling proteins such as THB1 would require an intermolecular electron transfer to accomplish this last step.

NO* plays a role as a signaling molecule in a variety of cellular processes, including several in the model organism Chlamydomonas reinhardtii, a close relative of C. eugametos. Specifically in the former, NO* inhibits the import of ammonium, nitrite, and nitrate, and also nitrate reductase activity. These observations support the idea that NO* can regulate the assimilation of nitrogen and direct attention to the possible involvement of globins in processes related to NO*.

Several TrHb genes have been identified within the nuclear genome of C. reinhardtii. Following the current UniProtKB annotations, these are designated THB1-THB4, but at least 10 putative genes may contain TrHb domains. Most C. reinhardtii TrHbs are hypothetical proteins. Exceptions are the products of “THB8”, the level of expression of which is increased >1000-fold under anoxic conditions, and “THB7” and “THB12”, which display mild upregulation under hypoxic conditions. These proteins have not been characterized fully, and the chemistry associated with them is unclear. Recently, the gene product of THB1 has been identified in in vivo work focused on intraflagellar transport (IFT) and the BBSome, an IFT cargo adaptor. In this case, as well, functional information is missing.

Knowledge of amino acid sequences and in vitro properties far exceeds physiologic data for the TrHbs of cyanobacteria and unicellular photosynthetic algae. In fact, it has been difficult to obtain biological information in part because these globins occur at relatively low levels and are generally not essential under most growth conditions. THB1 offers a rare opportunity to explore the properties of a TrHb and connect in vitro and in vivo information in a model organism. With the goals of testing specific functional hypotheses and relating in vivo and in vitro data, we report on the structure and reactivity of the purified recombinant protein and demonstrate the possible involvement of the protein in nitrogen metabolism.

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### MATERIALS AND METHODS

#### Chlamydomonas Cell Culture.
Strains CC-125, CC-1086, CC-1690, and CC-2453 were obtained through the Chlamydomonas Resource Center (University of Minnesota, St. Paul, MN). Strains g1 and bbs4-1 were from the Witman lab. Cells were maintained on tris acetate phosphate (TAP) medium agar plates until the cells were used. Liquid cell cultures were grown in Sager-Granick M medium, unless noted otherwise, at 20 °C under constant agitation and illuminated with cool white fluorescent light on a 14/10 (on/off) cycle to synchronize cell growth. For flagella isolation experiments, cells were aerated by being bubbled with 5% CO₂ and 95% air.

#### Antibody Production, Protein Analysis, and Gene Expression.
Polyclonal antibodies were prepared using the synthetic peptide AADTAPADSLYSRC, which corresponds to the first 13 amino acids of THB1 (excluding the initial Met) followed by a single cysteine for covalent attachment to the keyhole limpet hemocyanin carrier protein. Antibodies were raised in rabbit (Covance, Princeton, NJ), affinity purified using the same synthetic peptide, and stored frozen at a protein concentration of approximately 1.5 mg/mL in phosphate-buffered saline (PBS) until they were used.

Whole cell protein was extracted from actively growing cultures of C. reinhardtii by resuspending approximately 1 × 10⁶ cells/mL in 60 mM Tris-HCl (pH 7.0), 10% glycerol, 2% sodium dodecyl sulfate (SDS), and 50 mM tris(2-carboxyethyl)phosphine (TCEP, BondBreaker, Thermo Scientific), followed by boiling for 10 min and separating the protein on a 16.5% Tris-tricine SDS–polyacrylamide gel (Bio-Rad Laboratories, Hercules, CA). For silver staining, gels were fixed and stained according to manufacturer’s protocol (Silver Stain Plus Kit, Bio-Rad). Flagellar isolation was performed using the dibucaine method as previously described. In these experiments, whole cell samples were collected immediately prior to dibucaine treatment. After flagellar abscission, cell bodies were separated from flagella by centrifugation (3 min at 1150g). Concentrated whole flagella and corresponding whole cell and cell body proteins were separated on 4 to 15% Mini-PROTEAN TGX precast gels (Bio-Rad). Whole cell protein was extracted from actively growing cells using the RNeasy Plant RNA extraction kit with optional on-column DNA digest (Qiagen), and the quantity and purity of RNA were determined using a NanoDrop Spectrophotometer (Thermo Scientific). cDNA was synthesized from purified RNA using an iScript cDNA Synthesis kit (Bio-Rad) and analyzed using PrimeTime quantitative polymerase chain reaction (qPCR) probes (Table S1 of the Supporting Information) and 1X SsoFast Probes Supermix (Bio-Rad) on a Bio-Rad CFX Real Time Detection system using 40 cycles of 10 s, 95 °C melt steps and 20 s, 60 °C elongation steps per cycle. Normalized gene expression was calculated from biological duplicates performed in triplicate wells and normalized relative to the expression of the CBLP control gene using CFX Manager.
Recombinant Protein Production. The coding sequence for the THB1 gene was obtained from GenBank (EU095254.1) and used to create a synthetic gene that was inserted into the pJExpress414 plasmid after codon optimization for expression in a bacterial host system (DNA 2.0, Menlo Park, CA). Recombinant THB1 (below termed rTHB1) was obtained as described previously for Synechocystis sp. PCC 6803 GlnB.\(^\text{28}\) As in this case, overexpression results in the production of the apoprotein, which fractionates into inclusion bodies. Following urea solubilization and gel filtration chromatography, the holoprotein was prepared by addition of excess hemin (Sigma) to refolded apoprotein, further purified by anion exchange chromatography, and lyophilized for storage at −20 °C as necessary.

Mass spectrometry was performed on an Acquity/Xevo-G2 UPLC-MS instrument (Waters) and returned a molecular mass of 14564 Da for purified rTHB1, in excellent agreement with the mass of 14564.4 Da expected for the polypeptide lacking the initial methionine. The protein concentration was determined on a per-heme basis with an extinction coefficient of 8.066 × 10\(^4\) M\(^{-1}\) cm\(^{-1}\) at 630 nm in 0.1 M phosphate (pH 7.1) equilibrated with air (21% O\(_2\), 23% N\(_2\), 56% Ar) and 4% CO. When dithionite (DT) (Alfa Aesar, Ward Hill, MA) was not used, the protein was reduced with either the Fd/NADP+ reduction system or DT in 50 mM phosphate buffer (pH 7.1) equilibrated with air (~21% O\(_2\), 23% N\(_2\), 56% Ar) and 4% CO. When dithionite (DT) (Alfa Aesar, Ward Hill, MA) was not used, the protein was reduced with either the Fd/NADP+ reduction system or DT in 50 mM phosphate buffer (pH 7.1) equilibrated with air (~21% O\(_2\), 23% N\(_2\), 56% Ar) and 4% CO.

O\(_2\) Scavenging and Reduction Systems. When solutions needed to be scrubbed of dissolved O\(_2\), the protein was incubated, and data were collected in the presence of the coupled glucose oxidase/catalase oxygen-scavenging system (GODCAT).\(^\text{35}\) Final concentrations of the various components were 0.04 mg/mL bovine catalase, 2.8 mg/mL bovine catalase (Sigma), and 0.04 mg/mL glucose oxidase (Sigma). The pH titration of ferric and ferrous rTHB1 was obtained between pH 4.7 and 10.5. Spectra were recorded until saturation was observed. The ferric state was generated by incubation of the rTHB1 sample with GODCAT prior to the addition of DT. The ferrous cyano complex was obtained by reduction of the cyanometh complex with excess DT; likewise, the ferric NO\(_2\) adduct [Fe(II)−NO\(_2\)] was prepared by addition of a 5-fold excess of KCN. The ferric−NO\(_2\) adduct [Fe(II)−NO\(_2\)] was prepared by addition of a 5-fold excess of NO\(_2\) as released by 6-(2-hydroxy-1-methyl-2-nitrosohydrazino)-N-methyl-1-hexanamine (MAHMA-NONOate, Cayman Chemical, Ann Arbor, MI). In both cases, spectra were recorded until saturation was observed. The ferrous state was generated by incubation of the rTHB1 sample with GODCAT prior to the addition of DT. The ferrous cyanohem complex was obtained by reduction of the cyanometh complex with excess DT; likewise, the ferric NO\(_2\) adduct [Fe(II)−NO\(_2\)] was prepared by addition of a 5-fold excess of KCN. The ferric−NO\(_2\) adduct [Fe(II)−NO\(_2\)] was prepared by addition of a 5-fold excess of NO\(_2\) as released by 6-(2-hydroxy-1-methyl-2-nitrosohydrazino)-N-methyl-1-hexanamine (MAHMA-NONOate, Cayman Chemical, Ann Arbor, MI).

Nitric Oxide Dioxygenase and Griess Assays. The procedure used MAHMA-NONOate as the NO\(_2\) donor and the Fd/NADP+ reductase system for rTHB1 reduction. Specifically, ~10 μM rTHB1 samples were prepared from a concentrated rTHB1 stock in 100 mM phosphate buffer (pH 7.1) equilibrated with air (~21% O\(_2\), 23% N\(_2\), 56% Ar) and 4% CO. When dithionite (DT) (Alfa Aesar, Ward Hill, MA) was not used, the protein was reduced with either the Fd/NADP+ reduction system or DT in the presence of GODCAT, respectively.

Global fitting of significant vectors was performed with Savuka.\(^\text{35}\)

The pH titration of ferrous wild-type rTHB1 was obtained from separate samples for each pH value from pH 4 to 10.8. Each sample contained ~7 μM ferric rTHB1 in 100 mM buffer; different buffers were used depending on the desired pH (acetate/phosphate for pH 4.0−6.3, phosphate for pH 6.5−8, borax for pH 8.3−9.0, and glycine for pH 9.5−10.5). Spectra for each sample in the ferric state were collected prior to reduction to determine the concentration of the protein using the pH-dependent extinction coefficient determined in the ferric rTHB1 titration. Ferrous rTHB1 was then obtained by adding 2 mM DT to the sample, and the reduction was monitored as a function of time. Data points to form the titration curve were collected when the Soret band displayed maximal absorbance. The change in absorbance with time suggested damage to the protein caused by DT. The data may not correspond to a true equilibrium titration and are therefore less accurate than in the ferric state.
prepared with no rTHB1 present (“−THB1”) and was treated in the same manner as the sample containing rTHB1 (“+THB1”). Horse skeletal muscle Mb (Sigma) was subjected to an analogous procedure for the purpose of comparison. The kinetics of MAHMA-NONOate decomposition were determined in an independent experiment and yielded a half-life of $2 - 3$ min at pH 7 (not shown).

Nitrite was quantitated at the end of each experiment using the Griess assay.37 The NOD reaction mixtures were split into two fractions to which $4 \mu M$ FAD (final concentration) was added. To one of these samples was added A. niger nitrate reductase (NR, Sigma) to a final concentration of 0.075 unit/mL along with an NADPH regeneration system38 to convert nitrate into nitrite stoichiometrically. After $\sim 20$ min, both samples were treated with the Griess reagents (GR) as per the manufacturer’s instructions (Life Technologies, Grand Island, NY). Absorbance spectra were collected from 600 to 300 nm every minute until complete conversion was reached by the sample containing the largest amount of nitrite (20–50 min).

Nitrite concentrations were obtained from a calibration curve produced under the same conditions that were used for the NOD reaction and yielded an $\varepsilon_{520}$ of 26.9 $\mu M^{-1} cm^{-1}$ for the diazonium product. The same protocol was used to determine the amount of nitrate produced in NOD reactions in the presence of horse skeletal muscle Mb. Kinetic simulations were performed with KinTek Explorer.39

**UV–Visible Spectrophotometry.** Absorbance spectra were collected on either a Varian Cary-50 or an AVIV 14DS instrument. For extinction coefficient measurement, pyridine hemochrome spectra were acquired in triplicate, from 600 to 500 nm in 1 nm steps, using a 1 s averaging time per step. Ligand binding, NOD, and Griess assays were monitored using the Cary50 scanning kinetics mode. Optical spectra were collected using an averaging time of 0.1 s, from 700 to 300 nm, every 60 s, until sample equilibration was achieved.

**NMR Spectroscopy.** NMR data were acquired at 14.1 T with a Bruker Avance or Avance II spectrometer equipped with a cryoprobe. $^1$H chemical shifts were referenced to 2,2-dimethylsilapentane-5-sulfonic acid through the temperature-corrected $^1$H$_2$O line (4.76 ppm at 298 K). $^13$C and $^{15}$N chemical shifts were referenced indirectly in accordance with their respective $\Xi$ ratios.40 NMR data were processed using NMR Pipe 3.041 or TopSpin 2.1 (Bruker BioSpin, Rheinstetten, Germany). Analysis of NMR spectra was completed using Sparky 3.42

Unlabeled or uniformly $^{15}$N-labeled rTHB1 samples were exchanged from 50 mM Tris purification buffer into 300 $\mu M$ phosphate or 20 $\mu M$ phosphate (pH 7.5–8.0, 10% $^2$H$_2$O) and first examined in the ferric state in the absence of an exogenous ligand. Ferric wild-type, Y29F, and K53A rTHB1 samples ranged from 500 $\mu M$ to 3 mM heme in 270–600 $\mu L$ at pH 7.0–7.6. Initial $^1$H spectra were collected at either 298 or 308 K over a large spectral width to detect high-spin species.
increased population of the high-spin form of ferric rTHB1 was achieved by lowering the sample pH to 5.4 at 298 K.

To generate the ferrous state of wild-type, Y29F, and K53A rTHB1, a ~2 mM protein solution [25 mM borate (pH 9.1–9.5)] was placed in a glovebox under a continuous Ar (AirGas purity 4.8) stream. After approximately 10 min, the protein was reduced with 2.5 molar equiv of DT. The solution was transferred to an NMR Shigemi tube, and the tube was sealed with Parafilm. Samples produced in this manner remained completely reduced over the time scale of NMR data acquisition (days to weeks). Ferrous 15N-labeled WT and unlabeled Y29F and K53A rTHB1s were investigated by collecting one-dimensional (1D) 1H spectra with water presaturation. 1H–15N HSQC, 1H–15N histidine selective long-range HMQC data were collected on the wild-type protein, and homonuclear two-dimensional (2D) NOESY and DQF-COSY data were acquired on the Y29F variant as described previously.44

For electron self-exchange (ESE) measurements, a 1.8 mM sample of ferric wild-type [15N]rTHB1 in 250 mM borate (pH 9.2) and a 10% H2O/90% 1H2O mixture was incubated with GODCAT and purged with Ar as described above. A substoichiometric amount of DT was added, resulting in a mixture containing ferric and ferrous protein in an ~2:2:1 ratio. As estimated by the relative integrated intensities of resolved amide signals in relaxed 1H–15N HSQC spectra, the ferrous and ferric populations remained stable over the course of 24 h (final ferric:ferrous ratio of ~2:6:1).

The kinetics of ESE were estimated on a mixture of ferric and ferrous rTHB1 using a 1H–(N2)–1H ZZ exchange experiment. The sample was initially screened by comparing standard 15N shifts between ferrous and ferrous forms necessitated the use of 1H–(N2)–1H 2D ZZ exchange data.44,45 N2 ZZ mixing times were 452, 271, 701, 158, 565, 362, 1018, and 452 ms. 1H–15N heteronuclear cross-relaxation was minimized with 1.3 ms G3 1H inversion pulses every 10 ms during the N2 mixing time; these pulses were applied 2800 Hz downfield of the 1H2O line to limit the saturation of water magnetization, which was maintained at equilibrium during the ZZ period. Because of the low cross-peak intensity and sample variation with time, the data provide an estimate of the lower limit of the kinetics of ESE.

The cyanoform state of rTHB1 was prepared by incubation of ferric rTHB1 with a 1.5–6-fold molar excess of potassium cyanide for at least 30 min prior to data acquisition. Cyanomet wild-type rTHB1 samples were prepared at pH 7.5; protein concentrations ranged from 2 to 5.5 mM in either 10% or 99.9% H2O solutions. Data were acquired at 298 K. 1H homonuclear data (1D, NOESY, and DQF-COSY) and natural abundance 1H–13C HMOC data were collected on both wild-type and Y29F rTHB1 under similar conditions. In addition, wild-type rTHB1 was studied using 1H–1H TOCSY, 1H–13N HSQC, 1H–15N histidine selective long-range HMOC, three-dimensional (3D) NOESY-HSQC, and 3D TOCSY-HSQC experiments as described previously.46–48

RESULTS AND DISCUSSION

Figure 1A presents the primary structure of THB1, which has the features of a “group 1” TrHb (TrHb1 for short), one of three branches of the TrHb family.2 Of the few TrHb1s that have been studied, the closest relatives are also shown: C. eugametos LI6374 (CtrHb, 48% identity) and the cyanobacterial globins from Synechocystis sp. PCC 6803 (49% identity) and Synechococcus sp. PCC 7002 (46% identity). The three-dimensional structures of these proteins are available in the cyanoform [Protein Data Bank (PDB) entries 1DLY, 1S69, and 4L2M, respectively]. The Cx traces of the three structures superimpose with root-mean-square deviations of <0.8 Å over 90% of the chain. This similarity raises the expectation that the TrHb1 fold is robust and representative of THB1. The structure of cyanoform CtrHb with appropriate amino acid replacements (Figure 1B, prepared with SWISS-MODEL)49 was used to guide the analysis of the rTHB1 NMR data.

Purified rTHB1 appeared as a single band on silver-stained SDS–polyacrylamide gels (Figure 2A, lane 1). The heme-reconstituted protein is highly soluble in aqueous buffer and has a far-UV circular dichroism spectrum consistent with the helical content of CtrHb (Figure S1 of the Supporting Information). The high affinity of rTHB1 for heme and its reproducible behavior in solution suggest that this protein can serve as a good substitute for endogenous THB1. The characterization of the purified protein was initiated with optical spectroscopy to gain insight into heme coordination, association with various diatomic ligands, and stability to acid.

rTHB1 binds Diatomic Ligands. Figure 3A–D shows the absorption spectra of rTHB1 with and without the usual complement of diatomic ligands at neutral pH: O2, CO, and NO. In the ferrous state and CN− and NO− in the ferric state. Of note is the formation of the cyanide adduct of the ferrous state (Figure 3C) by DT reduction of cyanoform rTHB1. Ferrous cyano rTHB1, however, could not be obtained by addition of cyanide to reduced rTHB1. Absorbance maxima and extinction coefficients are listed in Table 1. Each spectrum is distinctive and can be used for the purpose of identifying species occurring in mixtures and enzymatic assays.

pH Dependence of Heme Axial Coordination in Ferric rTHB1. Figure 4 and Figure S2 of the Supporting Information illustrate the pH response of rTHB1 monitored via absorbance...
measurements in the absence of added ligand. Below pH 4.8, the Soret band experiences a sharp decrease in intensity (apparent pK_a of 3.72 ± 0.01 and a Hill coefficient of 3.45 ± 0.15, where the uncertainties are the errors of the fit); a shallower drop occurs at pH > 10.8 (apparent pK_a of 12.49 ± 0.01 and a Hill coefficient of 1.25 ± 0.03). Both acid and alkaline transitions (Figure S2 of the Supporting Information) are accompanied by spectral alterations between 400 and 700 nm characteristic of a change in heme coordination. The acid transition is likely coupled with protonation of the proximal histidine, the loss of heme from the protein cavity, and protein unfolding, whereas the nature of the alkaline transition is unclear as the spectral features at very high pH (broad absorption band at 588 nm, Soret maximum at 398 nm, and shoulder at 350 nm) do not match those of free hemin under the same conditions.

The spectral response in the range of pH 4.8–10.8 is shown for the visible region in Figure 4A. Figure 4B presents the pH titration curve described by the absorbance at 540 nm. The 430–700 nm data, when analyzed by singular-value decomposition,34 yield two principal abstract vectors. Global fitting35 returns an apparent pK_a of 6.48 ± 0.07 with a Hill coefficient of 0.86 ± 0.03 and indicates a transition involving one proton. However, imperfect isosbestic points (482, 522, and 598 nm) and the sloping baselines required to achieve a good fit (Figure 4B) hint at the participation of an additional low-population species or secondary ionization event.

The main spectral component between pH 4.8 and 6.5 is consistent with a high-spin, water-bound species.50 In contrast, the main component between pH 6.5 and 10.8 has low-spin character and is not readily recognized. Its appearance eliminates the possibility of a ferric hydroxy complex50 or a ferric pentacoordinate species51 and supports coordination of a protein residue to the iron on the distal side. This behavior parallels that of ferric CtrHb.52

**pH Dependence of Heme Axial Coordination in Ferrous rTHB1.** The response of ferrous rTHB1 to pH, shown in Figure 4C, is also reminiscent of that observed for CtrHb. Between pH 7 and 10, the protein spectrum is consistent with that of a six-coordinate low-spin complex.55 At

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**Table 1. Optical Properties of Various Wild-Type rTHB1 Complexes at pH 7.1**

<table>
<thead>
<tr>
<th>protein state</th>
<th>Soret maximum (nm)</th>
<th>α (nm)</th>
<th>β (nm)</th>
<th>CT^a (nm)</th>
<th>ε at Soret maximum (mM⁻¹ cm⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Fe(III) (pH 4.8)</td>
<td>408</td>
<td>503, 542 (sh)</td>
<td>630</td>
<td>144</td>
<td></td>
</tr>
<tr>
<td>Fe(III) (pH 7.1)</td>
<td>410</td>
<td>536, 568 (sh)</td>
<td>630 (sh)</td>
<td>125</td>
<td></td>
</tr>
<tr>
<td>Fe(III) (pH 10.8)</td>
<td>412</td>
<td>538, 570 (sh)</td>
<td>118</td>
<td></td>
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<tr>
<td>Fe(II)</td>
<td>426</td>
<td>529</td>
<td>559</td>
<td>195</td>
<td></td>
</tr>
<tr>
<td>Fe(II)--CN^-</td>
<td>417</td>
<td>546</td>
<td>104</td>
<td></td>
<td></td>
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<tr>
<td>Fe(II)--CN^-</td>
<td>432</td>
<td>534</td>
<td>564</td>
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<tr>
<td>Fe(II)--NO</td>
<td>421</td>
<td>535</td>
<td>570</td>
<td>136</td>
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<tr>
<td>Fe(II)--NO</td>
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<td>548</td>
<td>573</td>
<td>133</td>
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<tr>
<td>Fe(II)--O2</td>
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<td>581</td>
<td>119</td>
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<td>Fe(II)--CO</td>
<td>421</td>
<td>541</td>
<td>564</td>
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</table>

^aCharge transfer band.
acidic pH, the spectral features of a four-coordinate heme emerge with broad Soret and α/β bands suggesting loose association of the heme with the protein. The intermediate state, presumably a five-coordinate species with the proximal histidine as sole axial ligand, does not become highly populated under the explored conditions, although there is evidence of this state at neutral pH (Figure S3 of the Supporting Information). The apparent pK_a of 6.48 (see the text). Absorbance of the ferric protein at 540 nm as a function of pH. The solid line is the result of fitting the data according to a modified Henderson–Hasselbalch equation assuming a single ionization event \[ \text{pK}_a = 6.6. \]

Figure 4. pH titration of rTHB1. (A) The α and β bands of the ferric protein are shown between pH 10.8 and 4.8. The vertical arrows indicate the direction of the change as the pH is decreased. (B) Absorbance of the ferric protein at 540 nm as a function of pH. The line is the result of fitting the data according to a modified Henderson–Hasselbalch equation assuming a single ionization event \[ \text{pK}_a = 6.48 \] (see the text). (C) Absorbance of the ferrous protein at 559 nm as a function of pH.

Optical Data for the Endogenous Coordination on the Distal Side. Two plausible candidates for axial coordination on the distal side of the heme are the residues at position B10, as proposed in CtrHb, and E10, as in the two cyanobacterial GlbNs introduced in Figure 1A. In THB1, Tyr29 occupies the B10 position and Lys53 the E10 position. To determine the identity of the distal ligand, we first replaced Tyr29 (B10) with a phenylalanine. The Y29F variant binds heme tightly and yields optical spectra (not shown) similar to those of the wild-type protein. In addition, the pH response has transitions in the acidic, neutral, and alkaline regions that match closely those of the wild-type protein (Figure S2 of the Supporting Information). In particular, the transition near neutral pH occurs with an apparent pK_a of 6.82 ± 0.05 and a Hill coefficient of 0.97 ± 0.05, obtained by global fitting. The minor perturbations caused by the Y29F replacement suggest that, in ferric wild-type THB1, Tyr B10 is not an axial ligand to the iron.

In contrast, the K53A replacement causes pronounced changes in the optical spectrum and its pH response. At neutral pH, ferric K53A rTHB1 has a spectrum (Figure S)

Figure 5. Optical spectrum of K53A rTHB1 in the ferric state at four pH values and in the reduced state at pH 7.1. The ferric state undergoes an aquomet–hydroxymet transition at pH >8.

Figure 5. Optical spectrum of K53A rTHB1 in the ferric state at four pH values and in the reduced state at pH 7.1. The ferric state undergoes an aquomet–hydroxymet transition at pH >8.
predominant low-spin species (Figure 6C). In contrast, the K53A rTHB1 spectrum at neutral pH contains only the resonances of an aquomet complex (Figure 6D). As the pH is increased to 9.1, a broad spectrum consistent with mixtures of aquomet and hydroxymet forms is obtained (not shown).

Analysis of $^\text{1}H$−$^\text{13}C$ HMQC and DQF-COSY data identified the ring signals from Tyr29 and Phe29 in wild-type and Y29F rTHB1, respectively (Figure S5 of the Supporting Information). The moderate line widths and modest downfield shifts of Tyr29 ring protons are inconsistent with ligation to the heme iron. In neither case were any NOE contacts observed between the aromatic ring at position 29 and the heme cofactor. Signals from a potential sixth ligand were not immediately apparent in these spectra.

In the proximity of the iron, paramagnetic effects can broaden lines and render resonance identification difficult. These effects were attenuated by reduction to the ferrous state, which was conducted at high pH (9−9.5) to populate fully the diamagnetic complex according to the optical data. The water presaturation $^\text{1}H$ 1D spectrum of ferrous wild-type rTHB1 (Figure 7A) and ferrous Y29F rTHB1 (Figure 7B) at pH 9.5 (10% $^2\text{H}_2\text{O}/90\% \text{H}_2\text{O})$ and pH$^*$ 9.2 (90% $^2\text{H}_2\text{O}/10\% \text{H}_2\text{O}$), respectively, revealed a set of six single-proton resonances between −1.4 and −3.4 ppm. Such large upfield shifts are expected of nuclei directly above and below the porphyrin ring signals from Tyr29 and Phe29 in wild-type and Y29F rTHB1. A sample of uniformly $^{15}\text{N}$-labeled ferric wild-type rTHB1. A sample of uniformly $^{15}\text{N}$-labeled ferric wild-type rTHB1 in a 90% $\text{H}_2\text{O}/10\% \text{D}_2\text{O}$ mixture at 298 K.

Corresponding to two labile protons was detected at −7.8 ppm. Strong NOEs to the lysine aliphatic chain (Figure 8B) and to the heme $\alpha$ meso proton (not shown) and splitting into a doublet in $^{15}\text{N}$-labeled samples (Figure 7A) secure assignment to the $\text{NCH}_2$ of the lysine. The chemical shifts of the Lys53 side chain are listed in Table S3 of the Supporting Information. We conclude that the distal ligand in ferrous rTHB1 is Lys53. To the best of our knowledge, this is the first documented example of a His−Fe−Lys coordination scheme in a hemoglobin.

The pH titrations of ferric and ferrous wild-type rTHB1 (Figure 4) suggest that K53 is a heme ligand in the ferric state. To test this hypothesis, we sought to measure ESE in wild-type rTHB1. A sample of uniformly $^{15}\text{N}$-labeled ferric wild-type rTHB1 was prepared at pH 9.2 to favor the endogenous hexacoordinate state (Figure 4). Substoichiometric addition of DT resulted in a mixture of ferrous and ferric rTHB1. In fully relaxed $^\text{1}H$−$^{15}\text{N}$ HSQC spectra, the downfield-shifted amide NH signals from Asn87 (see below) report on the proportion of reduced and oxidized species. The mixture shown in Figure 9 originally contained approximately 31% reduced rTHB1. Little oxidation occurred during the course of data acquisition (28% reduced rTHB1 present at the end). Two-dimensional $^\text{1}H$−$^{15}\text{N}$ data were collected to take advantage of the resolved $^\text{1}H$ shifts and the apparent slow exchange between the two states. The observation of specific cross peaks connecting corresponding ferrous and ferric amide NH resonances is direct evidence that under these conditions, an electron transfer event interconverts the redox forms of rTHB1.

In favorable cases, measurement of the cross and diagonal peak intensities as a function of the $N_2$ mixing time provides data for determining the rate constant of exchange. The population drift and low signal-to-noise ratio prevented a detailed analysis of the exchange process, but an estimate could be obtained from the backbone amide cross peak of Asn87. An effective ESE rate constant was found to be 0.4 s$^{-1}$ at 1.8 mM total protein (apparent second-order rate constant of $\sim$0.2 mM$^{-1}$ s$^{-1}$). Although this number pertains to nonphysiological conditions, it indicates that rTHB1 can exchange electrons at a rate $\sim$1 order of magnitude slower than that of the bis-histidine GlnB formulations under similar conditions. ESE kinetics have been measured from $10^3$ to $10^5$ M$^{-1}$ s$^{-1}$ in a variety of heme proteins.

**Figure 6.** One-dimensional $^\text{1}H$ spectra of ferric rTHB1: (A) wild type at pH 7.6 (−9% high-spin), (B) wild type at pH 5.4 (intensity ×4 compared to trace A, −90% high-spin), (C) Y29F variant at pH 7.0 (intensity of downfield region X3, −40% high-spin), and (D) K53A variant at pH 7.4. Resonances marked a−d correspond to the 3-CH$_3$, 8-CH$_3$, and 2-vinyl $\beta$$_{meso}$ and $\beta$$_{ap}$ protons, respectively, from the low-spin wild-type and Y29F species. Data were recorded in a 10% $^2\text{H}_2\text{O}/90\% \text{H}_2\text{O}$ mixture at 298 K.

**Figure 7.** $^\text{1}H$ spectra of ferrous rTHB1. One-dimensional spectra of (A) $^{15}\text{N}$-labeled wild-type rTHB1 in a 90% $^2\text{H}_2\text{O}/10\% \text{H}_2\text{O}$ mixture (pH 9.5) with amide $^{15}\text{N}$ decoupling (120 ppm) during acquisition and (B) Y29F rTHB1 in a 10% $^2\text{H}_2\text{O}/90\% \text{H}_2\text{O}$ mixture (pH$^*$ 9.2). Heme meso, Asn87 NH, and resolved Lys53 signals are labeled.

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including cytochromes \( c \) (His-Met, \( 10^{-3} \text{ to } 10^{5} \text{ M}^{-1} \text{ s}^{-1} \)), cytochromes \( b_5 \) (His-His, \( 10^{1} \text{ to } 10^{4} \text{ M}^{-1} \text{ s}^{-1} \)), and aforementioned GlbNs (His-His, \( 10^{-1} \text{ to } 10^{3} \text{ M}^{-1} \text{ s}^{-1} \)). The rather slow kinetics for rTHB1 suggest that structural reorganization may limit the rate of electron transfer. Rapid lysine decoordination or solvent exchange dynamics may explain the slow but detectable ESE and be responsible for the broad line shape of the Lys53 \( \zeta \)H2. Overall, the data are consistent with Lys53 being the reversible distal ligand in both the ferric and ferrous states of rTHB1.

**Cyanomet rTHB1 Has a Distal H-Bond Network.** Upon addition of cyanide, the \(^1H\) NMR resonances of rTHB1 sharpen significantly (Figure 10A). This complex is stable and was chosen for the study of a representative species with a bound exogenous ligand. The number and intensity of hyperfine-shifted resonances indicate that two forms of the protein coexist in a 5:5:1 ratio. Heme signals from both forms are readily assigned with homonuclear NOESY, DQF-COSY, and TOCSY data, aided by natural abundance \(^1H\text{-}^{13}C\) HMQC spectra. Resonance attribution is performed on the basis of \(^13C\) shift, \(^1H\) intensity, and \(^1H\text{-}^{1H}\) scalar coupling patterns (Figure S7 of the Supporting Information contains the vinyl couplings). Assignment to specific peripheral substituents (Figure 1C) is achieved using intra-heme NOE connectivities. For example, the 1-CH3/2-vinyl and 3-CH3/4-vinyl pairs are distinguished by the observation of a small 8-CH3-to-1-CH3 effect. Figure S8 of the Supporting Information shows the heme group with relevant intra-heme NOEs. Table 2 lists the chemical shifts of the heme peripheral substituents. The heme methyl chemical shift pattern displayed by the minor form is consistent with heme orientational isomerism, as are NOEs (not shown).

The conformation of the distal pocket, where exogenous ligands bind and chemistry usually occurs, is an important determinant of reactivity. In cyanomet rTHB1, two exchangeable protons, not split into a doublet when the protein is uniformly labeled with \(^15N\), are detected at 27.0 ppm (major isomer) and 24.8 ppm (minor isomer) (Figure 10A). These efficiently relaxed protons have strong NOEs to resonances at 10.2 ppm (major isomer) and 9.68 ppm (minor isomer), themselves \(J\)-coupled to signals at 8.12 and 7.90 ppm, respectively. \(^13C\) data (Figure S9 of the Supporting Information) prescribe assignment to \(H\delta\) and \(He\) resonances of two (major and minor) tyrosine rings rotating rapidly about their \(C\gamma\text{-}C\zeta\) axis. Similar sets of \(O\gamma\)H and ring signals are detected in other cyanomet TrHb1 spectra and are characteristic of Tyr B10 as it forms a hydrogen bond to the axial cyanide. In rTHB1, this residue is Tyr29, as confirmed with partial sequential assignments secured with \(^15N\)-separated NOESY and TOCSY data. Ring-to-ring NOEs to the adjacent Phe28 indicate a geometry compatible with that of CtrHb (PDB entry 1DLY). Consistent with this, the spectrum of...
cyanomet Y29F rTHB1 (Figure 10B) does not contain any labile proton resonances downfield of 16 ppm, whereas the cyanomet K53A variant at pH 7.7. Select assignments are indicated on panels A and B as listed in Table 2 and Tables S4–S6 of the Supporting Information. Blue labels refer to the major heme orientational isomer, red labels to the minor isomer, and green labels to Phe29 within the Y29F variant.

The 1H−15N HSQC data of wild-type rTHB1 reveals two NH2 groups belonging to the major isomer and having 1H signals at 11.11 and 5.27 ppm (15N at 96.3 ppm) and 8.63 and −0.90 ppm (15N at 103.3 ppm). The first of these NH2 groups has NOEs to the rings of Phe28 and Tyr29 and is assigned to Gln54 (E11). The other is assigned to Gln50 (E7). A model of the distal hydrogen bond network to cyanide is illustrated in Figure S10A of the Supporting Information.

The sequence alignment of Figure 1A and analogy to the determined structures of cyanide-bound GlbN (PDB entry 4L2M) and CtrHb (PDB entry 1DLY) suggest that several residues should be in close contact with the heme group. Thus, an upfield-shifted valine with NOEs to the 3-CH3, 4-vinyl Hβ,cis and ring protons of Phe28 is assigned to Val94, in the G helix. Near the 1-CH3 group, the signals of a Leu and a Phe ring belong to Leu73 and Phe57, respectively. In contact with the 2-vinyl, Phe, Val, and Ala side chains are detected and identified as Phe91, Phe28, Val94, Val119, and Ala122. Tyr68, along with Phe57 and Leu73, is in contact with the 8-CH3 group, whereas the 5-CH3 group has NOEs to Phe41 and Phe42. A representative set of NOEs is shown in Figure S11 of the Supporting Information. All assignments are consistent with the observed scalar connectivities and 1H−15N-based sequential assignments. A model of the proximal side of the heme is illustrated in Figure S10B of the Supporting Information.

An additional conserved feature of TrHb1s is the formation of a side chain–main chain hydrogen bond between a histidine at the beginning of the G helix (Nδ acceptor) and an amide in the turn preceding it (NH donor). In rTHB1, this helix capping interaction involves Asn87 NH and His90 Nδ1 and is in large part responsible for the downfield shift of Asn87 NH (major 1H signal at 11.22 ppm; minor signal at 11.33 ppm). In the absence...
of an exogenous ligand, these resonances remain downfield-shifted and are useful markers of the redox state of the protein as mentioned above. Protein chemical shifts are listed in Table S4 of the Supporting Information.

Overall, the spectroscopic results confirm that cyanomet rTHB1 resembles in fold and heme pocket geometry the related Synechocystis GlbN (PDB entry 1S69), Synechococcus GlbN (PDB entry 4L2M), Mycobacterium tuberculosis TrHbN (PDB entry 1RTE), Tetrahymena pyriformis trHb (PDB entry 3AQ5), and C. eugametos CtrHb (PDB entry 1DLY). The distal hydrogen bond network that stabilizes exogenous ligands is present. Despite the conserved fold and the conserved Tyr B10, however, the proteins are significantly different in the absence of an exogenous ligand. At neutral pH, bis-histidine coordination of the heme iron through His F8 (proximal) and His E10 is observed in ferric and ferrous GlbNs.52,54,55 M. tuberculosis TrHbN and T. pyriformis TrHb bind water in the ferric state (aquomet form) and become five-coordinate species upon being reduced to the ferrous state.54,65 In C. eugametos CtrHb, Tyr B10 was proposed to be the sixth ligand to the iron,52 and in C. reinhardtii THB1, the distal ligand is Lys E10. These different coordination modes are expected to condition heme reactivity.

The structure of THB1 in the absence of exogenous ligand remains to be fully elucidated. The resemblance to the bis-histidine GlbNs of Figure 1A is expected because of the common E10 ligation. In GlbNs, Tyr B10 is not in the heme pocket, which would explain the small effect of the Y29F replacement on the acid–base transition observed at pH 6.5 in cyanomet rTHB1 (Figure S2 of the Supporting Information). This transition, attributed to the ionization equilibrium of Lys53, occurs ≈3.5 pH units lower than for a solvent-exposed lysine and indicates a free energy expenditure on the order of 20 kJ/mol, similar to the cost of ligating the proximal histidine at pH 366 but to be compensated for at neutral pH.

**rTHB1 Is Capable of Efficient NOD Activity.** NOD activity is a common function of globins that is often imputed to TrHbs. For example, M. tuberculosis TrHbN57 and T. pyriformis TrHb165 are both likely to serve in the capacity of pyriformis and His B10 was proposed to be the sixth ligand to the iron,52 and in C. reinhardtii THB1, the distal ligand is Lys E10. These different coordination modes are expected to condition heme reactivity.

The structure of THB1 in the absence of exogenous ligand remains to be fully elucidated. The resemblance to the bis-histidine GlbNs of Figure 1A is expected because of the common E10 ligation. In GlbNs, Tyr B10 is not in the heme pocket, which would explain the small effect of the Y29F replacement on the acid–base transition observed at pH 6.5 in cyanomet rTHB1 (Figure S2 of the Supporting Information). This transition, attributed to the ionization equilibrium of Lys53, occurs ≈3.5 pH units lower than for a solvent-exposed lysine and indicates a free energy expenditure on the order of 20 kJ/mol, similar to the cost of ligating the proximal histidine at pH 366 but to be compensated for at neutral pH.

To verify that rTHB1 turns over NO as the nitrosyl adduct of rTHB1 detected. The same experiment conducted with horse skeletal muscle Mb gave the profile shown in Figure S12 of the Supporting Information. To verify that rTHB1 turns over NO and forms nitrate, the end mixtures were subjected to the Griess assay.37 The Griess assay depends on the reaction of sulfanilic acid with nitrite to form an activated diazonium salt, ultimately producing the detected azo dye. Under conditions of excess Griess reagents, colorimetric product formation reports on nitrite concentration. In the absence of rTHB1, oxidation of NO by O2 generates detectable quantities of nitrite most likely through the NO + ½O2 reaction.64 In the presence of rTHB1, however, a negligible amount of nitrite is found in solution (Figure S13 of the Supporting Information). This indicates that an enzymatic

![Figure 11](dx.doi.org/10.1021/bi5005206i) Biochemistry 2014, 53, 4573–4589

Subsequent additions of MAHMA-NONOate lead to the same sawtooth behavior. The proportion of ferric state at the minimum of absorbance varies, with a tendency for an increase over the course of multiple NO injections. At concentrations of the NO donor capable of releasing 3 rather than 2 times the amount of protein, the ferric form binds NO, giving rise to the altered profile shown in Figure 11B. At no point was the ferrous–nitrosyl adduct of rTHB1 detected. The same experiment conducted with horse skeletal muscle Mb gave the profile shown in Figure S12 of the Supporting Information.

To verify that rTHB1 turns over NO and forms nitrate, the end mixtures were subjected to the Griess assay.37 The Griess assay depends on the reaction of sulfanilic acid with nitrite to form an activated diazonium salt, ultimately producing the detected azo dye. Under conditions of excess Griess reagents, colorimetric product formation reports on nitrite concentration. In the absence of rTHB1, oxidation of NO by O2 generates detectable quantities of nitrite most likely through the NO + ½O2 reaction.64 In the presence of rTHB1, however, a negligible amount of nitrite is found in solution (Figure S13 of the Supporting Information). This indicates that an enzymatic
reaction consumes NO* considerably faster than the enzyme-free processes. Addition of A. niger NR to convert nitrate to nitrite showed that in the rTHB1 sample, ~90% of NO* released by MAHMA-NONOate was oxidized to nitrate. A comparable yield was obtained with Mb.

The minimal set of reactions in the NOD assay includes (1) release of NO* by MAHMA-NONOate (first-order decay, measured independently), (2) reaction of NO* with rTHB1-O2 (eqs 2 and 3, considered irreversible), (3) reduction of ferric rTHB1 via reduced Fd, (4) Fd re-reduction, and (5) O2 binding by ferrous rTHB1 (reversible, measured independently in the absence of NO*). Although accurate rate constants for NOD activity cannot be obtained with this assay, it is possible to test the plausibility of the mechanism by simulating the kinetic traces corresponding to each NO* addition. Reproduction of the experimental data shown in Figure 11A can be achieved with a first phase dominated by the release of NO* by MAHMA-NONOate and a second phase limited by the rate of reduction of Fd by the NADPH system. At higher NO* concentrations (Figure 11B), reversible binding of NO* to ferric rTHB1 must also be included to account for the additional spectral feature.

The simplest interpretation of the data is that rTHB1 can process NO* efficiently in the test tube and that it undergoes no or little damage as it turns over the substrates. This supports the possibility that if O2 and NO* are simultaneously available to THB1 in the cell, NOD activity is likely. Multiple turnovers, however, require a reducing agent. Reductants such as NADPH and glutathione are not effective with the recombinant protein when used in reasonable concentrations (data not shown). The identity of the reducing agent in the cell, dedicated to THB1 or not, is not known.

The NOD assay was repeated with Y29F and K53A rTHB1. Qualitatively, the sawtooth pattern shown in Figure 11A is observed with both proteins (Figure 11C,D), and nitrate yields are comparable to those obtained with wild-type rTHB1 (Figure S13 of the Supporting Information). However, two differences are noted with the variants. First, the intensity of the Y29F rTHB1 spectrum decreases steadily throughout the experiment. We hypothesize as proposed by others65 that Tyr B10 stabilizes bound dioxygen and that its replacement with a phenylalanine exacerbates competing processes such as autoxidation (i.e., superoxide release), ultimately leading to heme damage. Second, the proportion of ferric protein reached after addition of MAHMA-NONOate to K53A rTHB1-O2 is higher than for the wild-type protein and Y29F variant, and the recovery to the oxy state is slower than for those two proteins. These observations are consistent with a coordinated Lys E10, once displaced from its axial position, to facilitating electron transfer from the reducing system. We also expect Lys E10, once displaced from its axial position, to engage in a salt bridge with a heme propionate and modulate iron accessibility.

**THB1 Levels Are Linked to the Nitrogen Source.** To complement the structural and chemical data, we pursued the in vivo characterization of THB1. To this end, we generated an anti-THB1 polyclonal antiserum. In a previous study,19 presence of THB1 was confirmed by the identification through mass spectrometry of two proteolytic fragments from the flagella of the C. reinhardtii bbs4-1 mutant. One of the peptides corresponds to an N-terminal extension to the truncated globin domain and most of the short α helix. The N-terminal extension is only a few residues long (Figure 1A), and analysis of the THB1 sequence with PredAlgo60, a program designed to reveal protein subcellular localization in green algae, does not suggest targeting to a specific organelle of the cell. On the basis of evidence that the N-terminal extension is expressed and unlikely to be important for protein targeting, this region of the protein was chosen for anti-THB1 antibody production.

Polyclonal antibodies raised against the N-terminal proteolytic fragment (red sequence in Figure 1A) recognize rTHB1 (Figure 2B, lane 1) and a protein within whole cell extracts of C. reinhardtii at the molecular weight that corresponds to full-length THB1 (Figure 2B, lane 2). The purified polyclonal antibodies appear to be highly specific; the only other detected band gives a weak response at a molecular weight slightly below 40000, seen only after prolonged overexposure (data not shown).

Given the assumption that THB1 interacts in some form with NO*, the native function of THB1 may well be linked to cellular activity that either creates NO* or is regulated by the molecule. Nitrogen metabolism is a potential target for THB1 as it is both regulated by NO17 and can produce NO* through a low-efficiency side reaction of NR acting on nitrite.70 Nitrogen usage in C. reinhardtii is complicated by the fact that not all laboratory strains metabolize nitrogen in the same way. Both CC-1690 and CC-125 are commonly used laboratory strains of C. reinhardtii; CC-1690 is assumed to be genetically identical to the original Sager 21 gr laboratory strain, but CC-125 carries mutations within both the NIT1 and NIT2 genes.71 The NIT1 gene encodes the only NR found within C. reinhardtii;72 it is responsible for reducing nitrate to nitrite via an NADPH-dependent reaction. NIT2 encodes a transcription factor required for expression of genes involved in nitrogen assimilation and is activated by nitrate.73 A cell with nonfunctional copies of either NIT1 or NIT2 is thus prevented from utilizing nitrate as a source for nitrogen. Figure 12A shows immunodetection of THB1 in protein extracts from whole cells grown in Sager-Granick M medium. THB1 is found in strain CC-1690 but is not detected in strain CC-125. The NIT1 and NIT2 mutations in the CC-125 strain can be separated using strains CC-1086 (containing only a mutation to the NIT2 gene) and CC-2453 (containing only a mutation to the NIT1 gene). As seen in Figure 12A, only strains with functional versions of the NIT2 gene express detectable levels of THB1 within the cell.

When nitrate is present in the growth medium, NIT2 induces expression of the nitrate assimilation pathway. Sager-Granick M medium uses 3.75 mM ammonium nitrate as the nitrogen source (i.e., equal amounts of ammonium and nitrate). When the M medium is modified to contain 7.5 mM ammonium chloride or 7.5 mM potassium nitrate, the amount of THB1 within C. reinhardtii strain CC-1690 varies (Figure 12B). When nitrate is absent from the medium, there is a marked depletion of THB1 protein from the cell. When cells are grown in original M medium and then placed in medium with only ammonium as a nitrogen source, the level of THB1 gradually decreases over the course of 48 h (Figure S14A of the Supporting Information). Conversely, when cells are grown in the presence of ammonium as the sole nitrogen source and then switched into the original M medium, detectable levels of THB1 appear within 6 h (Figure S14B of the Supporting Information).

The transcriptional regulation of the THB1 gene was followed using RNA isolated from these individual strains (Figure 12C). The level of gene expression for the THB1 gene was undetectable when either strain CC-12S or CC-1086 was analyzed; however, both CC-1690 and CC-2453 (containing...
interfering with phototaxis. These results also led to the suggestion that these proteins are continuously cycling through the flagella and that their export from the flagella is dependent on the BBSome.

The availability of an antibody to THB1 and the finding that THB1 expression could be regulated by the presence or absence of nitrate in the medium allowed us to reexamine the possibilities of a phototactic effect and a relation to the BBSome for THB1. To determine whether THB1 levels in the flagella are dependent on BBS4, we probed Western blots of isolated flagella from bbs4-1 and g1 with anti-THB1. As previously reported,19 THB1 was detected in the flagella of bbs4-1 but not g1 cells (Figure S15A of the Supporting Information). Surprisingly, THB1 was also either not present or present at exceedingly low levels in g1 whole cells or cell bodies lacking flagella. THB1 transcripts also were not detected in g1, similar to strains carrying nit2 mutations (Figure S15B of the Supporting Information). Although g1 was originally wild type for NIT2,24 our current g1 strain failed to grow with either nitrate or nitrite as the sole nitrogen source; in contrast, bbs4-1 was NIT2, as expected (Figure S15C of the Supporting Information). This suggests that the g1 strain maintained in the Witman lab and used in this study and that of Lechtreck et al.19 is a nit1 nit2 double mutant, most likely as a result of acquiring a spontaneous mutation in the NIT2 gene some time during the dozen or so years between generation of the bbs4-1 strain and the comparative analysis of the two strains by Lechtreck et al.19 The results indicate that the differences in THB1 levels in flagella of g1 versus bbs4-1 seen previously and again here simply reflect differences in THB1 expression. Importantly, flagella isolated from normally phototactic CC-1690 cells expressing both BBS4 and THB1 had levels of THB1 indistinguishable from that of bbs4-1 flagella (Figure 13). Therefore, a lack of BBS4 does not lead to an accumulation of functional NIT2 genes) possessed equivalent levels of the transcript. The CC-1690 strain reduced its level of THB1 transcripts when grown in a nitrate-free (7.5 mM ammonium) medium, in agreement with the depletion of the protein seen in the immunoblot (Figure 12A,B). In summary, these observations are consistent with induced expression of THB1 being dependent on the presence of nitrate.

The Presence of THB1 in Flagella Is Not Related to Flagellar BBS4 Levels. A previous study reported that THB1 and three other proteins, all identified by mass spectrometric analysis of bands excised from an SDS–polyacrylamide gel, were greatly enriched in flagella of the nonphototactic mutant bbs4-1 compared to its parent strain g1, which shows strong negative phototaxis.19 The bbs4-1 mutant was generated from strain g1 (nit1, NIT2) by insertional mutagenesis with a DNA fragment containing NIT1 as the selectable marker;74 the mutant is null for BBS4, which encodes a critical component of the BBSome, a multiprotein flagellar IFT cargo adapter.19 The apparent accumulation of THB1 and the other proteins in the flagella of bbs4-1 relative to the flagella of g1 raised the possibility that one or more of these proteins might be
THB1 in flagella, and the presence of THB1 in flagella is unlikely to explain the phototaxis defect in bbs4-1 cells.

Of the four proteins identified as being abnormally present in fractions of bbs4-1 versus g1 flagella, only THB1 was present in the aqueous phase enriched for flagellar matrix proteins; the rest, including phospholipase D (PLD), were present in the detergent phase enriched for membrane proteins. PLD has subsequently been shown to undergo a massive redistribution from the cell body to the flagella in the absence of BBS4; this redistribution is rapidly reversed when BBS4 is reintroduced into the flagella by fusing bbs4-1 cells with wild-type cells. The results of Figure 13 confirm that PLD accumulates in flagella of bbs4-1 cells but not in flagella of wild-type cells, in this case strain CC-1690. Therefore, PLD does cycle through the flagella, and export of PLD from the flagella is dependent on the BBSome. Such BBSome-mediated export may be a feature of many membrane-associated proteins that undergo such cycling. In contrast, because THB1 is a small soluble protein, it may diffuse freely through the barrier that separates the flagellar and cytosolic compartments; recent studies indicate that this barrier is permeable to soluble proteins of less than ∼50 kDa. Moreover, the fact that THB1 can enter the flagella strongly suggests that it is present in the cell cytoplasm and not strictly localized to a particular cell organelle. It may or may not have a specific function inside the flagellum.

**Possible THB1 Function.** Our results strongly suggest that the expression of the THB1 gene is not constitutive but rather linked to the source of nitrogen available to the cell. This expression pattern also resembles that of NR, which is known to be under the control of the NIT2 gene product. A minimal interpretation of the data holds that THB1 and NR expressions are linked because THB1 is required to eliminate the NO* released by NR. In addition to recycling the nitrogen to nitrate and lowering NO* levels temporarily, THB1 could also temper S-nitrosylation and other RNS reactions. A similar mitigation role has been proposed for GlbN in Synechococcus sp. PCC 7002. Like THB1, Synechococcus GlbN occurs at low levels in the cytoplasm, although unlike THB1, it is constitutively expressed. Challenging the Synechococcus cells with NO* or a high concentration of nitrate reveals that GlbN protects against ROS and RNS.

With its structural plasticity, its ability to bind ligands, and its NOD activity, THB1 may participate in downstream regulatory processes. Recent studies have indeed found multiple NO*-regulated pathways within *C. reinhardtii*, including copper stress response, hypoxia, and ammonium and nitrate transport. NO* regulation would not, however, be limited to THB1 because physiologic conditions exist under which exogenous NO* is present while synthesis of THB1 is dramatically curtailed. Another agent, possibly one of the other many putative hemoglobins, would be needed to manage the NO* load. In fact, potentially overlapping roles of all hemoglobins will complicate the identification of an unambiguous THB1 phenotype in future experiments with knock-down strains. The interrelation of NO*, RNS, and ROS chemistries in and out of the cell also presents special difficulties for the interpretation of results obtained under various challenges to cell growth.

**Conclusion.** We have shown that in THB1 the heme iron has two axial ligands: the proximal histidine in the F helix and a replaceable distal lysine (neutral Lys53) in the E helix. The structure of the His–Fe–Lys protein is expected to resemble that of GlbN in the bis-histidine state. The His–Fe–Lys scheme, like the His–Fe–His scheme, provides a means of enhancing electron transfer kinetics, affecting ligand association and dissociation rates, and influencing the reduction potential relative to a pentacoordinate counterpart. These effects can be harnessed to control NOD activity and other enzymatic reactions with requirements of redox cycling and substrate access and egress. *In vitro*, rTHB1 is an efficient NO* dioxygenase, and the link to nitrogen metabolism, specifically expression of NR, suggests that a similar activity occurs in the cell. Further investigation will establish the distinct advantages and disadvantages of the His–Fe–Lys scheme under various cellular conditions of ROS load, RNS load, and pH.

Hemoglobins first appeared more than 3 billion years ago in a biosphere substantially different from ours. Without significant atmospheric oxygen and with a host of highly reduced molecules, stringent control of redox chemistry must have been essential to the earliest forms of life. Hbs have been retained since those early days and have remained within the genetic toolbox of organisms across the entire spectrum of life. Many of these Hbs have been proposed to function in roles related to reactive nitrogen usage. *C. reinhardtii* is a well-studied model organism, and exploring the function of TrHbs within this alga will harness a wealth of knowledge for defining the role of these proteins. The study of THB1 presented here constitutes a first step toward a comprehensive investigation of photosynthetic microbial Hbs.

**ASSOCIATED CONTENT**

 Supporting Information

qPCR probes, 1H chemical shifts of the heme group in ferrous Y29F rTHB1, 1H chemical shifts of the axial lysine (K53) in ferrous Y29F rTHB1, 1H chemical shifts and NOEs of selected heme pocket residues in cyanomet wild-type rTHB1, 1H chemical shifts of the heme group in cyanomet Y29F rTHB1, 1H chemical shifts and NOEs of selected heme pocket residues in cyanomet Y29F rTHB1, circular dichroism spectrum of ferric rTHB1, pH titration of ferric wild-type and Y29F rTHB1, optical spectrum of ferrous wild-type rTHB1 at three pH values, pH titration of ferric wild-type and K53A rTHB1, aromatic region of the 1H–13C HMQC spectrum of ferric wild-type and Y29F rTHB1, identification of the K53 N=H signal in ferrous wild-type rTHB1, H signature of the heme vinyl groups in wild-type cyanomet rTHB1, diagram of intra-heme NOEs in wild-type cyanomet rTHB1, model of the heme pocket of cyanomet THB1, key heme pocket NOEs in cyanomet rTHB1, NOD assay of myoglobin, representative Griess assay results, THB1 levels over time as a function of nitrogen source, and determination of THB1 expression and Nit1 and Nit2 phenotypes for g1 and bbs4-1. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

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ABBREVIATIONS

1D, one-dimensional; 2D, two-dimensional; 3D, three-dimensional; BBS, Bardet-Biedl syndrome; CB, cell body; CtrlhB, heme domain (residues 44–164) of C. elegans LI637; DQF-COSY, double-quantum-filtered correlation spectroscopy; DT, sodium dithionite; ESE, electron self-exchange; ET, electron transfer; FAD, flavin adenine dinucleotide; Fd, spinach ferredoxin; GODCAT, glucose oxidase/catalase dioxygen scavenging system; GR, Griess reagents; Hb, hemoglobin; HMQC, heteronuclear multiple-quantum coherence; HSQC, heteronuclear single-quantum coherence; IFT, intraflagellar transport; MAHMA-NONate, 6-(2-hydroxy-1-methyl-2-nitrosohydratio)-N-methyl-1-hexanamine; Mb, myoglobin; NOD, nitric oxide dioxygenase; NOESY, nuclear Overhauser spectroscopy; NR, nitrate reductase; N2O, longitudinal 15N magnetization; PAGE, polyacrylamide gel electrophoresis; pH*, pH uncorrected for isotope effects; PLD, phospholipase D; RNS, reactive nitrogen species; ROS, reactive oxygen species; rTHB1, recombinant THB1; SDS, sodium dodecyl sulfate; TAP, tris acetate phosphate; TCEP, tris(2-carboxyethyl)-phosphine; THB1, product of the THB1 gene of C. reinhardtii; TOCSY, total correlation spectroscopy; TE, Tris-EDTA; TrHb, truncated Hb; TrHb1, group I TrHb; WC, whole cell; ZZ, mixing of longitudinal magnetization.

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